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Question 1 Part A What

is a possible mole-mole factor for the following equation?
 $5 \text{ C} + 2 \text{ SO}_2 \rightarrow \text{CS}_2 + 4 \text{ CO}$

Hint 1.

Mole-mole factors come from the coefficients of a balanced chemical equation. See Section 9.2 (Page 252).

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Problem 9.113 Problem

9.113 Assume that you

have 1.95 g of

nitroglycerin in a 605.5

mL steel container at

20.0 °C and 1.00 atm

pressure. An explosion

occurs, raising the

temperature of the

container and its

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Mole-mole factors come

from the coefficients of

a balanced chemical

equation. See Section

9.2 (Page 252) .

ANSWER: Correct

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Question 2 Part A Iron reacts with oxygen in the air to produce rust (Fe₂O₃) Page 4/25

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flashcards, games and other study tools. On the periodic table, one of the columns within the main group elements; a family or group of elements exhibits similar chemical properties. (3 Mastering chemistry answers chapter 3. 2).

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